

***“Influence of waterquality,  
EC and pH on sustainability in  
relation to chemical cropprotection  
in horticulture” .....***

"Hariduse juured on kibedad,  
aga vili on magus" ...

Aristoteles

## Market trends (in Europe)



- The **demand** (*customer awareness*) for environmentally friendly production has increased
- Regular loss of products because of this reason (*no sales*)
- Margins for growers are under pressure = production costs control and higher cultivation reliability are necessary.
- Trade and consumer demand higher quality and stronger product



# How to fight diseases in our crops:

## 1. What does fungi need to start:

- Spores or other sources of contamination.
- Susceptible plants
- High humidity and right temperature

## 2. Which action is the best to do:

- Remove all spores and avoid contamination
- Use the right chemicals/treatment against fungi
- ***Increase plant quality***
- ***Control the climate***

Success %

0

20

50

30

} 80% !!



# Why is sustainability a future need?

- It becomes very difficult to register new chemicals, due to environmental reasons.
  - The number of chemicals goes down every year.
  - Development new ones limited and very expensive.
- A sustainable production is WHO target for future development of agriculture world wide. (World Health Organisation)
- Society opposition against chemicals to avoid damage to nature, water, bees, and bio-diversity.

# GET AWAY FROM GUT FEELING !!!

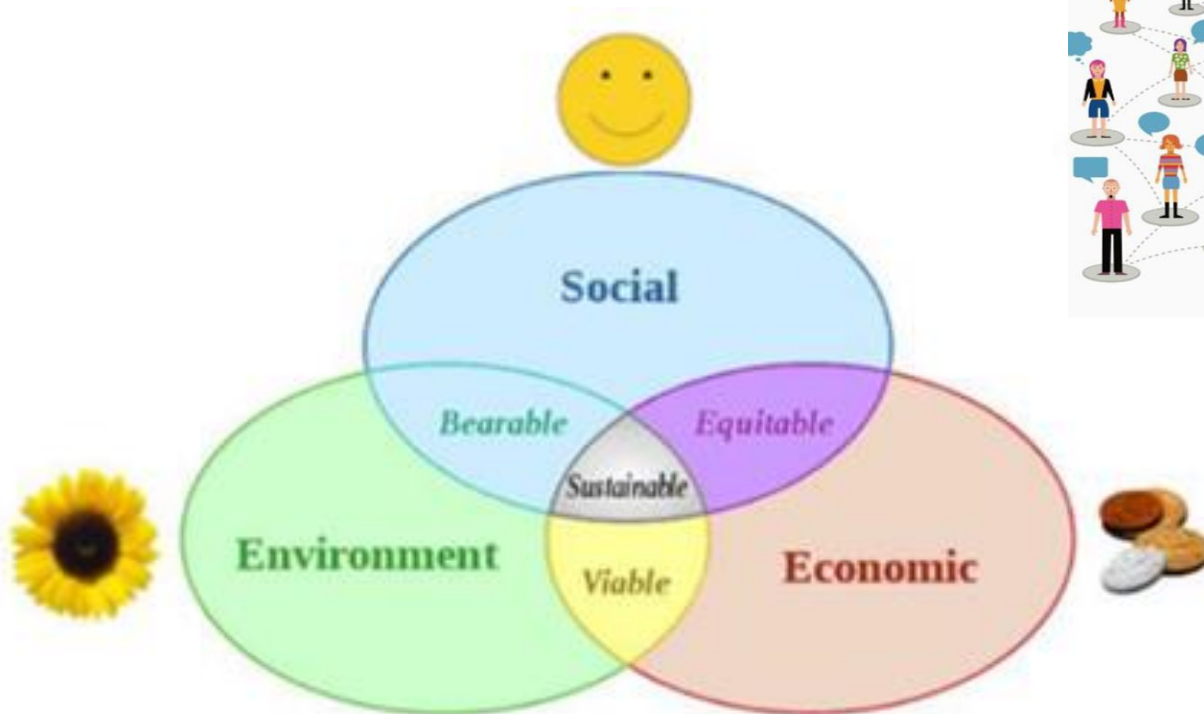


- Instead of defending the **relative** quality of plants,
- we'll need **to invest** significant resources in the creation of repeatable, scientific growing processes and procedures
- that will focus on **objective data points** rather than subjective “gut feelings”

# Sustainability

*Most issues are combinations of social, environmental and economic.*

*If you create a solution that considers all three, then you are being sustainable and have achieved a win – win - win*



*Viable = realistic*  
*Equitable = fair*  
*Bearable = tolerable*

# What should **YOU** do ?

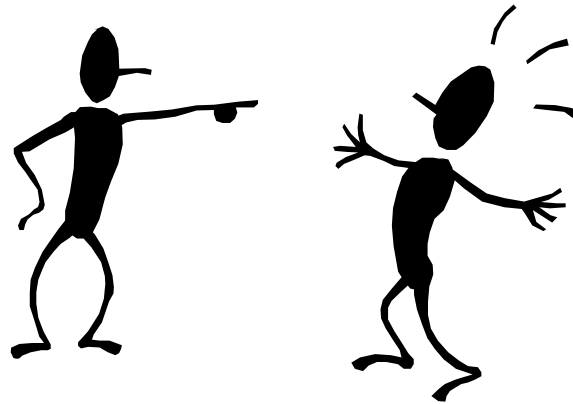
## Focus

*what's the problem*



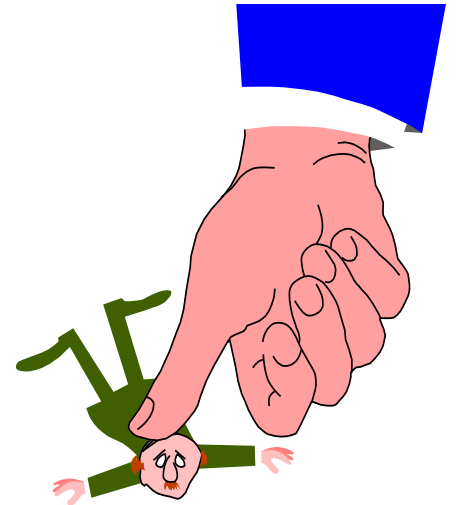
## Choose

*which to solve first*



## Solve

*how to do it*



# These topics we are going to highlight today

- **Substrate** Future peat-free compositions
- **Greenhouse climate** Some attention points
- **Light** Amount and spectrum
- **Water** Quality, Stability, Availability,
- **Nutrition/Fertilization** Composition, EC levels, Growth inhibition
- **New developments** Resilience cultivation, New developments

# Substrate

## Future peat-free compositions

These partners (University/NGO's) are involved in the development and building peat-free future



### Goal:

- Clear targets for **2025, 2030, and 2050** to increase the use of renewable resources.
- Investment in innovation and research to develop alternative raw materials and growing media.
- Responsible peat use with mandatory certification through the *RPP* label.
- Active consumer education and market transparency.

# Strategy of participating partners

- Step 1. Increasing the supply of **renewable** raw materials
- Step 2. Pre-processing and **stabilization**
- Step 3. Sanitation and **Plant Health**
- Step 4. Food **Safety** and Environmental Impact
- Step 5. Basic **Knowledge** and Application
- Step 6. Application-Oriented Knowledge **Development**
- Step 7. **Regulations** and specific applications

## Biochar.

*Residual material released during the production of methane from organic raw materials at high temperatures and oxygen-free conditions (temperatures >600°C).*



# Basic requirements substrate

- reliable supplier
- constant quality
- don't save on the price
- add a standard nutrition + pH buffer
- good structure till end of production
- air filled porosity
- waterholding capacity
- disease free
- pH 5,5 - 6,5
- EC +/- 0.6 – 0.8
- packaging / storage



50 cents/litre

30 cents/litre

## Current substrate parts

- White and black peat
- Cocopeat, coco fibre
- Composted Bark / wood fibre

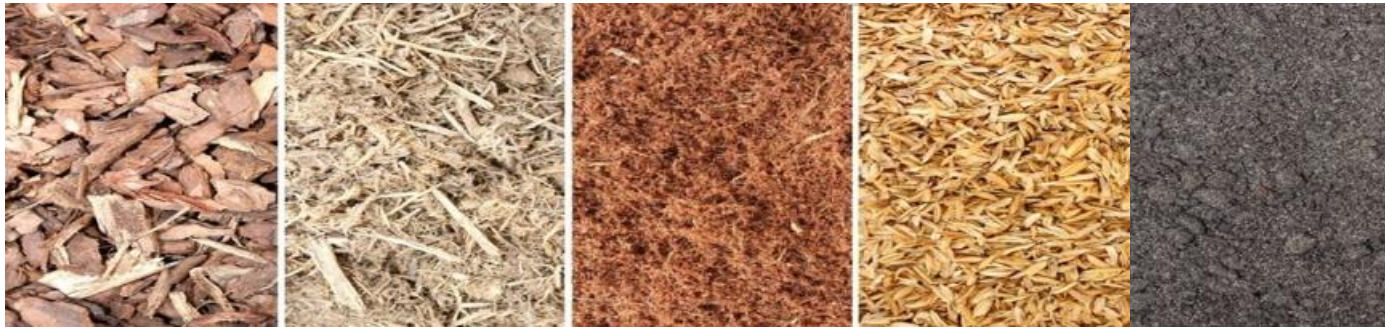


<https://www.rhp.nl/en/new-growing-media>

## Future renewable peat-free substrate parts

- Biochar
- Woodfibre
- Expanded clay
- Bark
- Cocopeat
- Rice husk
- woody waste
- poorly degradable straw
- easily degradable fibers
- Bagasse (By-product of sugar cane)
- By purpose produced green products
- Etc....

# Examples of alternative peat-free products.



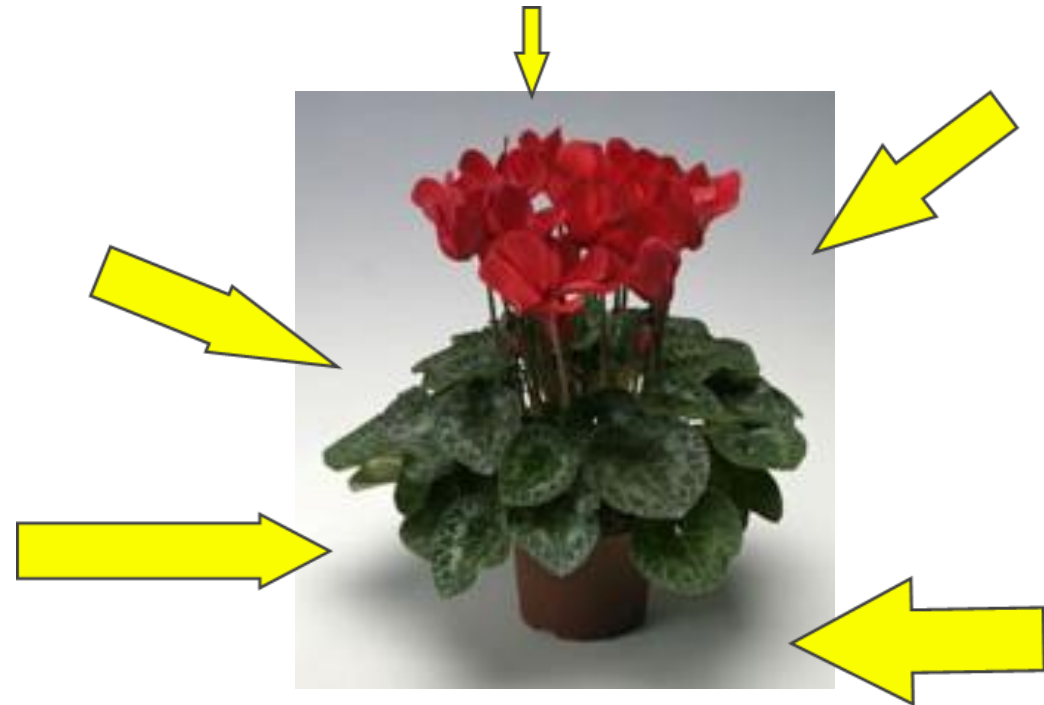
bark woodfibre cocopeat rice husks biochar



# Greenhouse *climate* related topics

- Temperature
- Air humidity
- Light
- Radiation
- Air circulation
- CO<sub>2</sub> = Carbon dioxide
- Pesticides, growth retardants
- (air pollution)

Climate is everything in the surrounding of the plant



# Botrytis, firsts sign of a bad climate

- Infection takes place by *wounds or direct penetration* of *soft tissue*. (this tissue being either very young or old).
- Germination of spores: *4-25°C and high humidity*
- The amount of water needed for germination and infection of Botrytis is a *relative humidity of 94%* during at least *8 hours* or an *8 hour period of free water* on the leaves.
- Lesions appear within *2 to 3 days at 4°C* and *within 24 hours at 25°C*.



# Summary of Climate rules

- don't give too much water. *(look at daylength, temp. light intensity)*
- try to grow as regularly as possible. *(like flying, or driving)*
- carry off always (high) humidity. *(keep windows open)*
- prevent big shakes in temp. RH and radiation. *(light intensity)*
- use heating system in time. *(even in August nights)*
- be aware of day and night differences. *(spring and autumn)*
- control climate between the plants. *(is far higher than average)*
- react proactive on weather changes. *(be informed about weather forecast)*
- avoid problems by taking action immediately. *(don't postpone)*
- ***Ventilate!!*** Also in winter. ***(RH and CO<sub>2</sub>)***

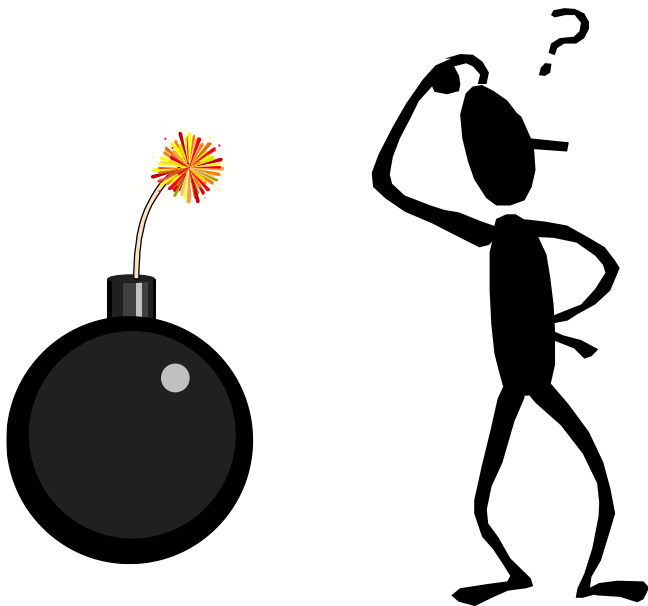
Remember !!

A

# Bad climate

is the most important reason of all possible problems !!

and one of the first aspects  
that must be solved as soon  
as possible.



# Light

Light spectrum has a significant impact on a wide variety of plant processes including photosynthesis and photomorphogenesis which in turn impact plant productivity.

- **Your assumption??**

- In Estonia easily the minimum factor in winter?

- **Negative for Estonia**

- Daylength, low sun position to horizon

- **Positive for Estonia**

- Frost days is regular clear sky
- Snow means light reflection

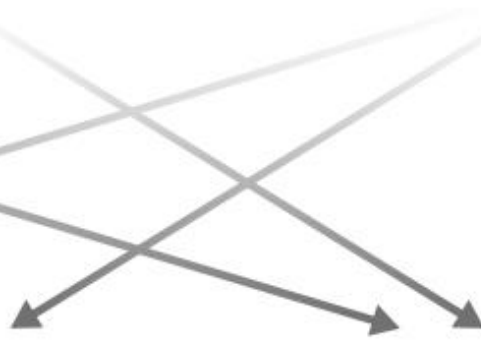
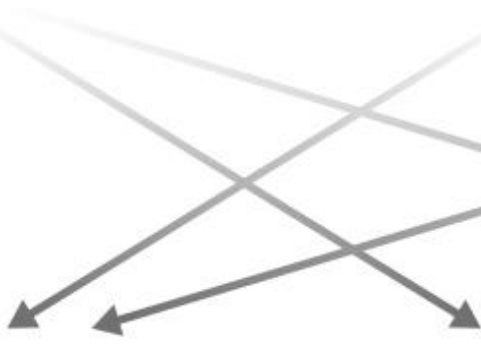
# Three dimensions of light for plants



Light quantity  
(intensity)

Light quality  
(spectrum)

Light duration  
(photoperiod)

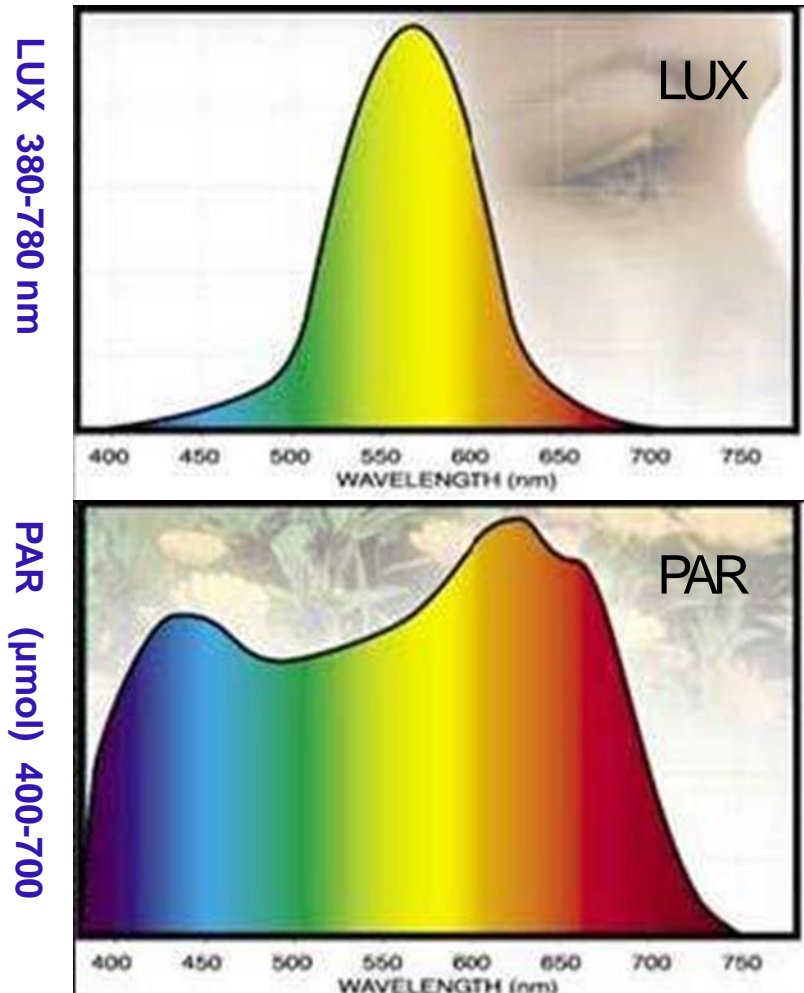


Plant biomass

Morphology

Flowering

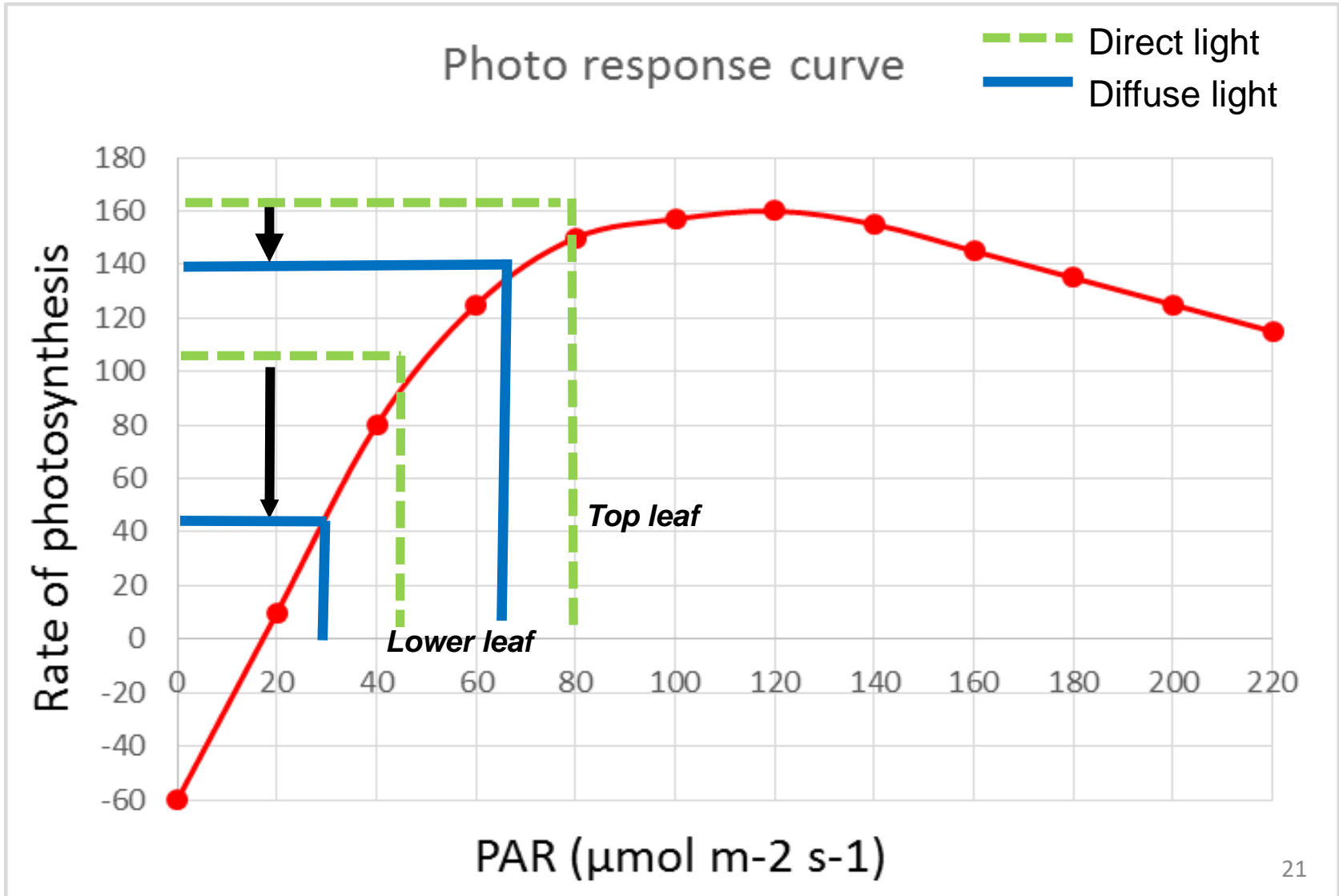
# Light response difference human eye and plants.



- Light HPS-lamps (high pressure sodium) is yellow-orange:
  - highest sensitivity of human eye
- Light in horticulture:
  - $\mu\text{mol}/\text{m}^2/\text{s}$  is the only reliable,
  - Lux unfit for use!
- Use a PAR-meter!
  - (photosynthetic active radiation)

# Difference in leaf activity = assimilation speed

## Fotosynthese with direct and diffuse light

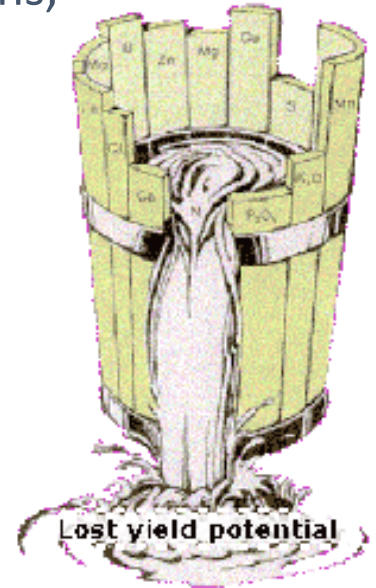


# Water

## Most important ingredient of the plant.

The next questions need to be answered:

- Do I know the quality of the water?
- Is there a big chance on change? Seasonal, weather conditions, environment, pollution?
- Is it recent information? Test it once a year!
- Does it meet my quality demands? And what are they?
- If not, do I have alternatives?
- Can I adapt the fertilizer accordingly?
- Do I need some help?



# Water

The most important factor in plant production

In case the quality is out of line next problems can occur

- Destroyed soil structure
- Cause direct salt damage to leaves and roots
- Individual ion toxicities or deficiencies
- Wrong pH which decreases fertilizer uptake
- Introduction and spread of fungi and bacteria
- Stunting, chlorosis, poor development

*Advice: rain water should be the first choice*

# Eurofins Holland laboratories water analysis result

Results		analysis	guideline	converted results	
	pH	7,8	5,5-7,5		
mS/cm 25°C	EC	0,2	<0,5		
Cations mmol/l	NH <sub>4</sub>	< 0,1	<0,3	< 1,9	ppm
	K	< 0,1	<1,0	< 4,0	ppm
	Na	0,6	<1,5	14	ppm
	Ca	0,7	<1,0	28	ppm
	Mg	< 0,1	<1,0	< 2,5	ppm
Anions mmol/l	NO <sub>3</sub>	0,5	<2,0	31	ppm
	Cl	0,6	<1,5	21	ppm
	S	< 0,1	<1,0	< 3,3	ppm
	HCO <sub>3</sub>	1,0	<3,0	61	ppm
	P	< 0,04	<0,20	< 1,3	ppm
Micro-nutrients µmol/l	Fe	0,2	<10	11	ppb
	Mn	< 0,1	<3	< 5,5	ppb
	Zn	0,5	<2	33	ppb
	B	3,0	<10	32	ppb
	Cu	< 0,1	<1,0	< 6,4	ppb
	Mo	< 0,1	<0,8	< 9,6	ppb
mmol/l	Si	0,04	<1,5	1,1	ppm

- All important elements
- Analysis in mmol and µmol per liter
- Converted into ppm = mg per liter
- A guideline for the risky levels

Converted results: ppm = mg/l and ppb = µg/l.

# Rainwater collection

## Advantage: free water, high quality

- Collecting equipment
- Gutters, pipes,
- Basin, tanks, pumps,
- Plastic on slopes with basin in case no sufficient tunnels
- Yearly amount of rain + melted snow
- Average Estonia 750 mm total \* +/- 50% = 400 mm / year
- Example: 400 mm = 400 ltr / m<sup>2</sup>
- **1000 m<sup>2</sup> greenhouse** \* 400 ltr / m<sup>2</sup> = 400.000 ltr/year means **400 m<sup>3</sup> in total per year.**



**400**

# Waterquality examples Estonian growers

EC	pH	NH4	K	Na	Ca	Mg	NO3	Cl	SO4	HCO3	PO4	Si	Fe	Mn	Zn	B	Cu	Mo
0.50	8.00	0.10	0.30	1.80	1.10	0.60	0.40	1.70	0.30	1.70	0.04	0.11	1.60	1.30	0.40	11.00	0.10	0.10
0.50	7.50	0.10	0.30	1.00	1.30	0.60	0.40	0.80	0.80	1.70	0.04	0.04	32.00	2.50	0.30	2.70	0.10	0.10
0.80	7.10	0.10	0.30	0.50	3.70	0.90	0.30	0.40	0.90	5.90	0.04	0.13	0.90	0.40	0.10	9.70	0.10	0.10
0.60	7.30	0.10	0.30	0.80	1.90	0.80	0.10	0.60	0.60	4.00	0.04	0.09	0.40	0.10	0.10	20.00	0.10	0.10
0.40	7.40	0.10	0.20	0.60	1.10	0.70	0.10	0.60	0.20	2.20	0.04	0.06	0.80	0.10	0.40	1.00	0.20	0.10
0.80	7.30	0.10	0.30	0.90	3.10	1.30	0.60	0.80	0.50	4.90	0.04	0.16	0.80	0.20	1.20	1.00	0.20	0.10
0.29	7.66	0.01	0.10	0.09	1.22	0.36	0.01	0.07	0.05	3.11	0.00	0.00	0.54	2.18	0.31	4.63	0.09	0.05
0.70	7.17	0.04	0.10	0.00	3.14	0.70	0.03	0.00	0.51	6.36	0.00	0.00	1.07	364.03	96.36	4.72	64.52	6.05
0.90	7.50	0.10	0.40	1.00	3.20	1.00	0.40	1.20	1.70	4.00	0.04	0.08	0.20	1.50	0.50	2.60	0.10	0.10

- All of them has **too hard** water, and need adjustments ( $\text{HCO}_3 > 0.5 - < 1.0$  mmol/ltr)
- pH has no direct link with hardness. ( $\text{HCO}_3$ )
- Huge differences per grower means different fertilization schedule for the same crops!

# Main impact of minor elements:

Fe	Mn	Zn	B	Cu	Mo	Si
1,60	1,30	0,40	11,00	0,10	0,10	0,11
32,00	2,50	0,30	2,70	0,10	0,10	0,04
0,90	0,40	0,10	9,70	0,10	0,10	0,13
0,40	0,10	0,10	20,00	0,10	0,10	0,09
0,30	0,10	0,10	3,50	0,10	0,10	0,10
0,80	0,10	0,40	1,00	0,20	0,10	0,06
0,80	0,20	1,20	1,00	0,20	0,10	0,16
0,54	2,18	0,31	4,63	0,09	0,05	0,00

## Too less / too much

- Mn - = chlorosis young leaves like Fe.  
+ = less Ca transport
- Zn - = malformation of leaves  
+ = antagonism with Fe.
- B - = dying tips, roots and leaf edges  
+ = necrotic dots in leaf edge
- Cu - = complete yellowing young leaves  
+ = dwarfs, lack of Iron
- Mo - = N-deficiency  
+ = gold-yellowing of shoots

# Problems of a wrong pH

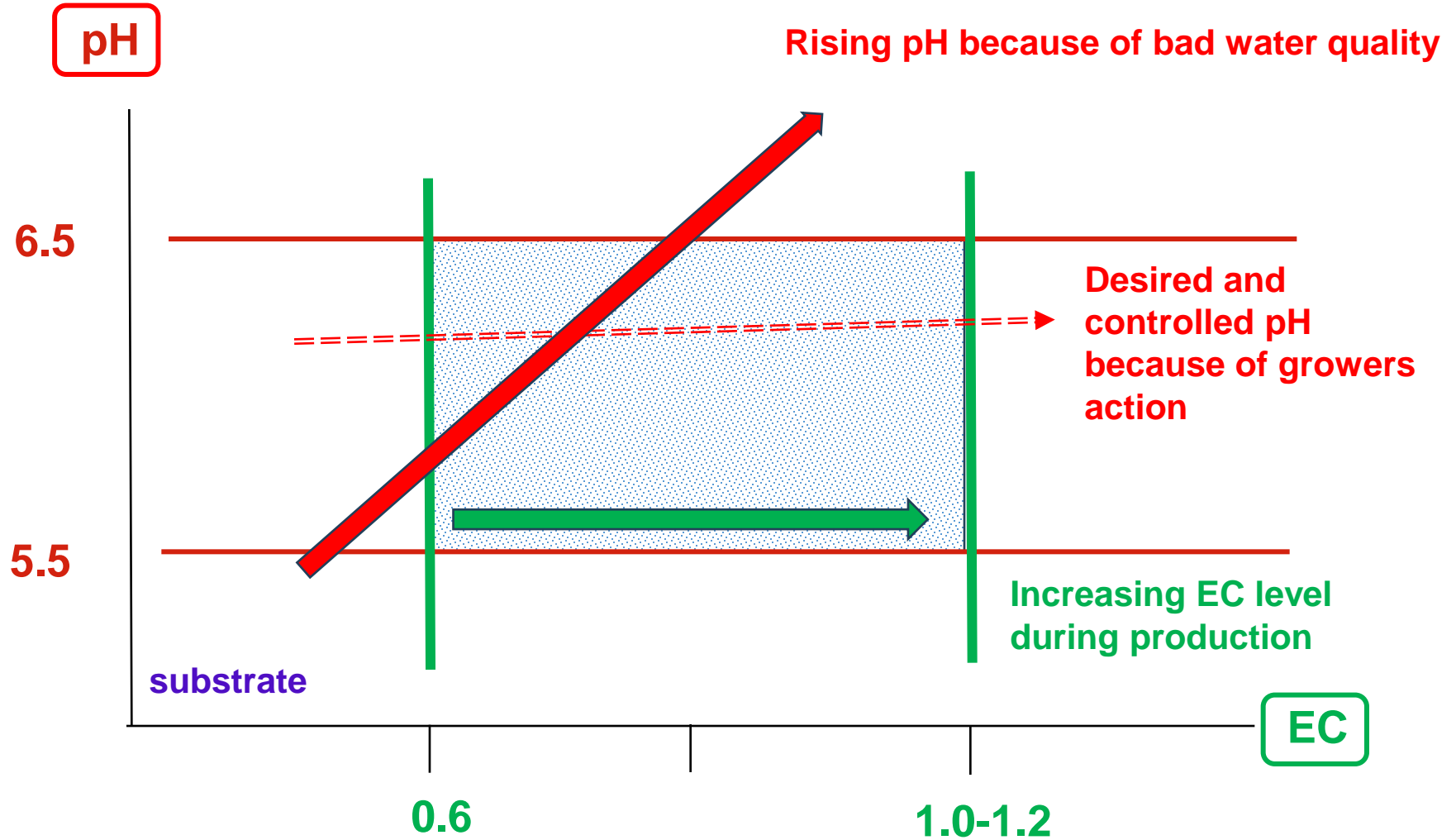
- In general the different elements are too much or too less soluble, what gives a big imbalance in availability of them.
  - = bad development and bad quality
- **Too low** = bad uptake of Phosphorus and Calcium
- **Too high** = bad uptake of minor elements as Mn, B, Fe
- *pH changes are going up or down **rather slow**, so pH level need constant attention and action.*



# Overview of solubility of elements at different pH

<p><b>Excess</b></p> <p>Manganese (Mn) Iron (Fe) Boron (B) Copper (Cu) Zinc (Zn) Sodium (Na) Nitrogen (NH<sub>4</sub>)</p> <p><b>Deficient</b></p> <p>Calcium (Ca) Magnesium (Mg) Phosphorus (P) Potassium (K) Sulfur (S) Molybdenum (Mo)</p>	<p>Readily available</p>	<p><b>Excess</b></p> <p>Calcium (Ca) Nitrogen (NH<sub>4</sub>)</p> <p><b>Deficient</b></p> <p>Manganese (Mn) Iron (Fe) Boron (B) Copper (Cu) Zinc (Zn) Magnesium (Mg) Phosphorus (P)</p>
<p>pH range</p>	<p>5. 5</p>	<p>6. 5</p>

# Optimal pH values (your situation?)



## Problems of a too high pH



pH = yellowing = Fe lack  
What makes the difference?

**Most sensitive crops:**  
Petunia and Calibrachoa

# Problems of a too high pH



➤ pH problem in Petunia seedlings

and

Gerbera youngplants

The most simple  
option to  
increase water  
quality...??



***Dilution!!***

***With rainwater quality***



**MACRO NUTRIENTS** or major elements which are required by the plant in fairly **large amounts**.

- **6 MINERAL NUTRIENTS**

- N, Nitrogen
- P, Phosphorus
- K, Potassium
- Ca, Calcium
- Mg, Magnesium
- S, Sulphur

- **3 NON-MINERAL NUTRIENTS**

- C, Carbon
- H, Hydrogen
- O, Oxygen

**MICRO NUTRIENTS** or minor elements which are required by the plant in relatively **small amounts**.

- **6 MICRO NUTRIENTS**

- Fe, Iron
- Zn, Zinc
- Mn, Manganese
- B, Boron
- Mo, Molybdenum
- Cu, Copper
  
- (Cl, Chlorine)

# EC 1,048 and still yellowing and no shoots

*What does this mean?*



# Low nitrogen treatment



- Result: weak, hungry
- sensitive for diseases
- too early flowering
- no branching

= actual way of thinking !!



## Too less food, no quality



*What about shelf life you think ?*

## Importance of the right fertilization !!



**Lack of Boron**



# Lack of Calcium in fruits and salad



Blossom End Rot



Cork or dod edge in apple



Edge in lettuce

# Allocation of elements 1:

## Easy/mobile: K, N, S, Mg

- They can be relocated from older plant parts
- Shortages first visible in the older / lower parts of the plant.



NO<sub>3</sub>



K



NO<sub>3</sub>

## Allocation of elements 2:

### Difficult/immobile: Ca, Fe

- Fertilisers are fixed in the plant and can't be relocated
- Shortages first visible in the youngest parts of the plant.



Ca



Fe



# EC = Electric Conductivity from the total diluted salts

**Does the EC inform us about the fertilizer level of the substrate?**

**NO**

- Attention: the EC is the value of the TOTAL diluted elements
- It can be only NaCl, so it say nothing about the chemical composition.

**Yes**

- When you have a complete analysis from the water or the substrate
- Or you know which fertilisers has been given the last weeks.

# Possibilities of *Growth Regulation*

- Choice of genetic material. (variety)
- Light = distance, space, additional light
- Amount of water
- **Amount of fertilizer** (*EC level*)
- **Balance of elements.** (*LOW PHOSPHATE*)
- Average growth temperature
- DIF and quick temperature drop
- Chemicals (*Bonzi, Alar, CCC*)

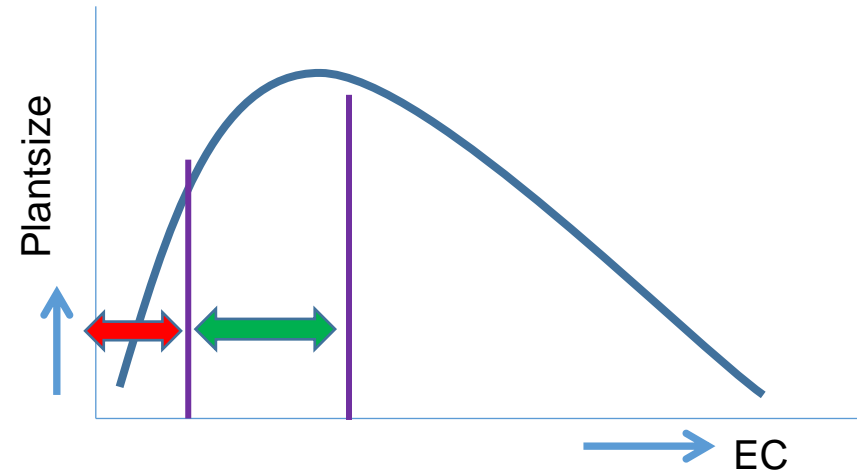


Or a  
combination  
of these

# Balans water stress ↔ fertilizers

## Optimal EC depends on:

- Crop
- Development phase
- Climate
- Day/night rythm (daylength, assimilation capacity)
- High radiation → more cooling = more evaporation needed → EC ↓
- Too high EC → difficult calcium transport
- Target values substrate > 0.8 -1.2 EC in the substrate at sales
- In most cases the drip EC should be almost doubled



## EC differences trial: size and color



Salvia



Zinnia

# Cyclamen EC differences



Too low

Okay

Too much

Results of the right EC use....

NO PGR's.



## A- and B- tank system

- Why are all fertilisers shared over an A-tank and an B-tank?

Because Calcium and Sulphate forms insoluble gypsum at high concentrations ( $\text{CaSO}_4$ )

- Which elements (ions) are not allowed together, and why not?

So keep Calcium always apart from  $\text{SO}_4$  and  $\text{PO}_4$



# Example A- and B-tank system on rainwater

Drip EC: 2.0	Grower: Sittig
Tank size (Litres): 1000	Country: South Africa
Concentration: 100%	Culture: Pansy etc.
Water from well: 0%	Soil analysis: No
Ureum: 0%	Advice: standard whole culture
Potas. Phosphite Y/N:	Water: <b>Rainwater</b>
P2O5 +/- : 100%	Schedule: F 12 - Group 3 generative
SO4 +/- : 100%	
NH4 /- : 100%	

	1000 litre A	Tank composition	1000 litre B
Calcium Nitrate (sol)	80.6 Kg.	Nitric Acid 60%	--- L.
Ammoniumnitrate (liq) 18.0%	7.4 L.	Phosphoric Acid 59%	--- L.
Potassiumnitrate	11.3 Kg	Potassiumnitrate	38.6 Kg.
Nitric Acid 60%	--- L.	Mono Potassium Phosphate	14.5 Kg.
Magnesiumnitrate (liq)	--- L.	Magnesium Sulphate	26.3 Kg.
Ureum	--- Kg.	Potassium Sulphate	23.2 Kg.
Calcium Chloride	--- Kg.		
	101.0 Kg.		102.7 Kg.
Fe. DTPA (sol) 11.6%	705 Gr.	Manganese Sulphate (Sol)	85 Gr.
		Zinc Sulphate (sol)	81 Gr.
<i>% Calcium Nitrate kg's 40%</i>		Borax (sol)	143 Gr.
<i>% Magnesium Sulfaat kg's 13%</i>		Copper Sulphate 25% (sol)	13 Gr.
<i>% Potassium Sulfaat kg's 11%</i>		Sodium Molybdate (sol)	12 Gr.
<i>HCO3 buffer mmol 0.00</i>			

Balance:	N	P	K	Mg	Ca
	1	0.37	2.0	0.21	1.0

# Example A- and B-tank system on groundwater

Drip EC: 2.0	Grower: Sittig
Tank size (Litres): 1000	Country: South Africa
Concentration: 100%	Culture: Pansy etc.
Water from well: 100%	Soil analysis: No
Ureum: 0%	Advice: standard whole culture
Potas. Phosphite Y/N:	Water: Groundwater
P2O5 +/- : 100%	Schedule: F 12 - Group 3 generative
SO4 +/- : 100%	
NH4 /- : 100%	

	1000 litre A	Tank composition	1000 litre B
Calcium Nitrate (sol)	58.0 Kg.	Nitric Acid 60%	12.8 L.
Ammoniumnitrate (liq) 18.0%	10.1 L.	Phosphoric Acid 59%	--- L.
Potassiumnitrate	18.9 Kg	Potassiumnitrate	31.9 Kg.
Nitric Acid 60%	4.3 L.	Mono Potassium Phosphate	14.5 Kg.
Magnesiumnitrate (liq)	--- L.	Magnesium Sulphate	12.1 Kg.
Ureum	--- Kg.	Potassium Sulphate	19.3 Kg.
Calcium Chloride	--- Kg.		
	94.7 Kg.		93.7 Kg.
Fe. DTPA (sol) 11.6%	686 Gr.	Manganese Sulphate (Sol)	81 Gr.
		Zinc Sulphate (sol)	78 Gr.
<i>% Calcium Nitrate kg's</i>	31%	Borax (sol)	--- Gr.
<i>% Magnesium Sulfaat kg's</i>	6%	Copper Sulphate 25% (sol)	10 Gr.
<i>% Potassium Sulfaat kg's</i>	10%	Sodium Molybdate (sol)	10 Gr.
<i>HCO3 buffer mmol</i>	1.00		

Balance:	N	P	K	Mg	Ca
	1	0.35	1.9	0.20	1.0

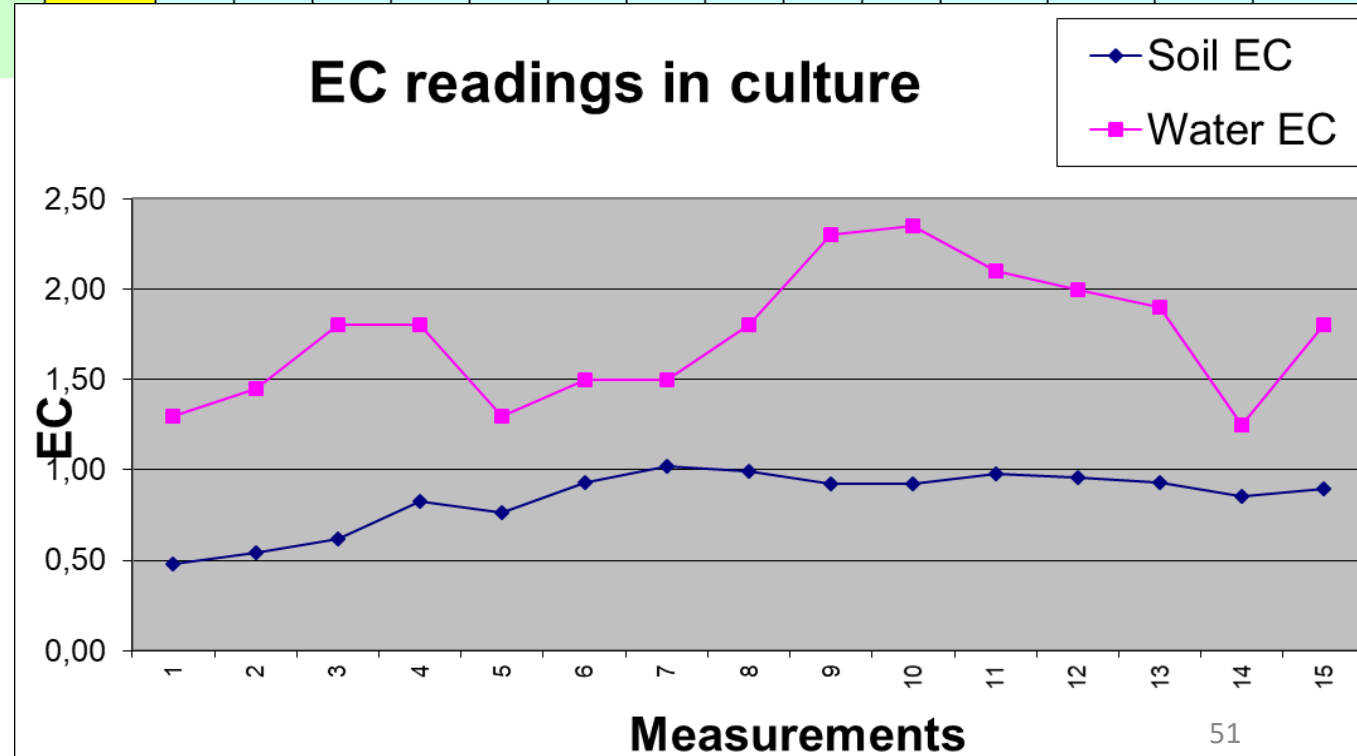
# History and calculation is basis of today action. EC and pH

## Substraat EC eigen metingen

Week nummer	Datum	Gewas	Oppot week	Geplande leverweek	Druppel Water EC	EC pot 1	EC pot 2	EC pot 3	EC pot 4	EC pot 5	EC pot 6	EC pot 7	EC pot 8	EC pot 9	EC pot 10	Hoogste meting	Laagste meting	Aantal potten	Gemiddelde pot EC
27	4-Jul-14	Cyclamen	18	34	1.50	0.62	0.64	0.67	0.63	0.63	0.64	0.72	0.51	0.66	0.64	0.72	0.51	8	0.64
28	11-Jul-14	"	18	34	1.50	0.71	0.57	0.82	0.63	0.54	0.58	0.83	0.76	0.72	0.66	0.83	0.54	8	0.68
29	18-Jul-14	"	18	34	1.80	0.89	0.61	0.75	0.74	0.83	0.61	0.79	0.73	0.80	0.51	0.89	0.51	8	0.73
30	25-Jul-14	"	18	34	1.80	0.90	0.70	0.89	0.61	0.75	0.74	0.83	0.61	0.79	1.00	1.00	0.61	8	0.78
31	1-Aug-14	"	18	34	2.00	0.92	0.89	0.67	0.89	0.61	0.75	0.74	0.83	0.61	0.79	0.92	0.61	8	0.77
32	8-Aug-14	"	18	34	2.00	0.78	0.89	0.61	0.75	0.74	0.83	0.61	0.79	1.20	0.88	1.20	0.61	8	0.78
33	15-Aug-14	"	18	34	2.30	0.70	0.78	0.99	0.88	0.90	0.98	0.79	0.88	0.90	1.30	1.30	0.70	8	0.89
34	22-Aug-14	"	18	34	2.30	0.90	0.78	0.89	0.83	1.04	1.08	0.76	0.88	1.19	1.50	1.50	0.76	8	0.95
35	29-Aug-14	"	18	34	2.30	0.80	0.78	0.88	0.83	1.19	1.20	0.79	0.91	1.34	1.76	1.76	0.78	8	0.99
36	5-Sep-14	"	18	34	2.00	0.86	0.78	0.70	0.82	1.33	1.31	0.81	0.93	1.48	1.00	1.48	0.70	8	0.98
37		"																	
38																			
39																			
40																			

*Look at drip EC differences compared with substrate EC.*

***What's going on here?***



## **Resilience** cultivation, means future license to grow.??

Ability of enough and effective chemicals for crop protection will become a bigger problem in the near future due to legislation and social acceptance. It is a must to prepare now already for that future.

Piisavate ja tõhusate taimekaitsekemikaalide olemasolu muutub lähitulevikus seadusandluse ja sotsiaalse aktsepteerimise tõttu suuremaks probleemiks.

Selleks on vaja juba praegu valmistuda

# Summary Plant disease management strategies

## Follow the Traditional Principles of Plant Disease Control

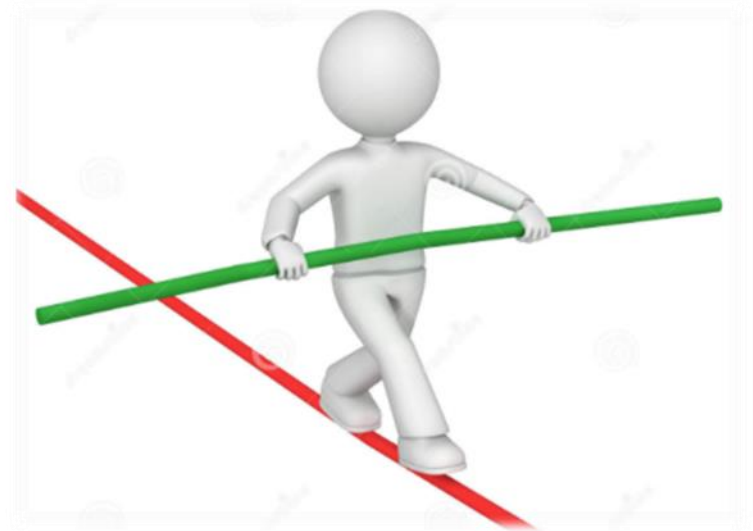
- **Avoidance** — prevent disease by selecting a crop, a season or a site where the environment is not favorable for infection.
- **Resistance** — utilize cultivars that are resistant to or tolerant to infection.
- **Daily action** — **make plants happy** with balanced fertilization, pH and EC control
- **Exclusion** — prevent the introduction of disease/pest. (control incoming material)
- **Protection** — prevent infection by means of a barrier to infection. (insect mesh)
- **Eradication** — eliminate, destroy, or inactivate the enemy. (and affected plants)



Finally...

*Edu ei ole lõplik, ebaõnnestumine ei ole saatuslik.*

*Oluline on julgus jätkata.*



Winston Churchill